

Physico Chemical Analysis Of Drinking Water Of Pedavegi Mandal Westgodavari District, Andhrapradesh, India

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Abstract: Water not only essential for human existence and also play vital role in agriculture, industries, household, recreation etc. Water which is safe for human consumption is called potable water. Suitability of bore water for human consumption in some of the rural areas of Pedavegi mandal are analyzed by measuring various physico chemical parameters such as P^H , turbidity, EC, TDS, TH, TA, chloride, fluoride, nitrite, sulphate, sodium, potassium, calcium, magnesium, iron, DO, BOD and COD. The results were compared by WHO¹ standards of water quality. The result reveals that all the water samples are suitable for drinking purpose.

Keywords: Bore well water, Physico chemical parameters, Water quality,

I. Introduction

Water is essential for the maintenance of life, without which life cannot flourish. It is well known that human survival depends upon use of uncontaminated and clean water for drinking and other purposes. People residing in this study area are forced to use bore water for their domestic and drinking consumption. The use of fertilizers and pesticides manure, lime, septic tank, refuse dump etc are the main sources of bore water pollution². The quality of the ground water is of great importance in determining the suitability for drinking, domestic and industrial purpose. The quality of the water varies from place to place, with the depth of the water table, from season to season and also the extent dissolution of dissolved solids present in it.^{3,4} Water must be tested with different physico – chemical parameters. Selection of parameters for testing of water is solely depends upon for what purpose we going to use that water and what extent we need its quality and purity. Hence an attempt was made to identify the potential areas for the sources of drinking water.

Study Area:-Present study deals with study of various physico chemical parameters of drinking water collected from bore wells located in the study area of pedavegi mandal, west Godavari district. The West Godavari district consists of 46 mandals, out of which 24 mandals are upland areas and 22 mandals are delta areas. One upland mandal, pedavegi is selected for the analysis of various parameters of bore water samples seasonally. Pedavegi is located in between 17.18358 to 17.31718 North latitude and 81.25935 to 81.45478 East longitudes.

Figure 1-view of AP in india



Figure 2-view of districts in AP



Figure 3-view of mandals in West Godavari District

Figure 4-view of pedavegi mandal



Sample Collection:- The water samples are collected from different bore wells located in Pedavegi mandal of west Godavari district. The collected water samples are labeled. The samples are collected in a clean polythene bottle as per standard procedures recommended in APHA (1991) ⁵. Samples were brought to the laboratory for analysis of various physico chemical parameters.

Methodology:- It is very important and essential to analyze the water before using for drinking, domestic, agriculture and industrial use. To assess the quality, water must be analyzed for various physico-chemical parameters such as P^H, turbidity, EC, TDS, TH, TA, chloride, fluoride, nitrite, sulphate, sodium, potassium, calcium, magnesium, iron, DO, BOD and COD. The results were compared by WHO standards for drinking water.

P^H: P^H of the water samples are measured by Eutech – 2700 P^H meter.

Turbidity:- Turbidity of the water samples are measured by Nepheloturbidimeter.- Systronics -132.

Electrical conductivity (E.C):- Electrical conductivities of the water samples are measured by Systronics -304 E C meter.

Total dissolved solids (TDS):- Total dissolved solids are measured by Evaporation methods - Gravimetrically.

Total hardness, Calcium, Magnesium (TH, Ca, Mg):- Total hardness, Calcium, Magnesium are measured Complex metrically 6,7 by EDTA titration method.

Total Alkalinity (TA):- Total alkalinity of the water samples are measured volumetrically by Titration with standard acid solution.

Sodium and Potassium: - Sodium and Potassium of the water samples are measured by Flame photometer.-127.

Dissolved Oxygen (D.O), Biological Oxygen demand (B.O.D).

Dissolved Oxygen (D.O), Biological oxygen demand (B.O.D) of the water samples are measured by some standard methods.

Flourides, Chlorides, Sulphates:- Flourides, Chlorides, Sulphates of water samples are measured by ion selectivity meter- Eutech 2700.

Table 1. Specifications for drinking water

S.No	Parameter	W H O Standards
1	p ^H :	6.5 to 8.5
2	Total Dissolved solids(T.D.S).	500 mg/L to 2000 mg/L
3	Total Alkalinity(T.A)	200 mg/L to 600 mg/L
4	Total hardness(T.H)	200 mg/L to 600 mg/L
5	Calcium (Ca)	75 mg/L to 150 mg/L
6	Magnisium (Mg)	35 mg/L to 70 mg/L
7	Sulphates	200 mg/L to 400 mg/L
8	Chlorides	250 mg/L to 1000 mg/L
9	Flourides	1.0 mg/L to 1.5 mg/L
10	Sodium	Up to 200 mg/L
11	Potassium	Up to 12 mg/L
12	Dissolved Oxygen (D.O)	4 mg/l to 6 mg/L
13	Biological oxygen Demand (B.O.D)	6 mg/L to 10 mg/L

Table 2.1 Pedavegi Mandal -- Rainy season

S a m p l e N o	p H	E C	T D S	Tu rbi dit y	Al kal init y	Ha rd nes s	So di um	Pot assi um	Ca lci um	Ma gne siu m	I ro n	Ch lor ide	Flu ori de	Ni tri te	Su lp ha te	Ph osp hat e	D O	C O D	B O D
01	8:7 1:5 1S2	27 70 20	17 72 8	00	578	250	140	7.1	12.02	53.59	0	523	0.862	0	101.5	0	4.0	8.0	1.2
02	8:7 2:9 1S4	17 10 04	77 00 4	00	390	310	75	4.8	26.05	59.68	0	480	0.601	0	54.5	0	4.4	17.6	2.8
03	8:7 3:5 1S4	17 60 24	17 20 4	00	366	265	66	5.5	46.09	36.54	0	396	0.591	0	32.5	0	4.0	8.0	2.0
04	8:7 4:5 1S2	17 40 20	18 99 6	00	436	260	61	2.6	38.07	40.19	0	221	0.788	0	41.5	0	4.0	9.6	2.4
05	8:7 4:5 2S8	13 30 80	83 33 2	00	366	160	75	2.3	44.08	12.18	0	109	0.593	0	50.6	0	4.4	17.6	2.4
06	8:7 5:9 1S0	17 00 00	16 40 4	00	354	220	85	3.2	22.04	40.19	0	85.7	1.16	0	38.2	0	4.0	22.4	2.4
07	8:7 6:7 1S1	17 20 68	17 06 8	00	440	320	143	5.6	26.05	62.18	0	274	0.871	0	50.3	0	3.2	32.6	1.1
08	8:7 7:4 1S3	27 30 72	14 07	00	520	225	102	8.8	46.09	26.79	0	594	0.778	0	74.1	0	3.6	25.6	1.1
09	8:7 8:8 1S8	17 40 80	18 09 6	00	384	180	55	11.5	34.06	23.14	0	256	0.616	0	66.5	0	3.6	36.8	1.1
10	8:7 9:5 1S2	17 00 40	16 40	00	370	300	67	3.1	24.04	3.65	0	219	1.199	0	37.7	0	4.8	32.8	2.8
11	8:7 9:8 2S6	17 40 60	18 09 6	00	364	195	63	4.7	30.06	29.23	0	139	1.21	0	56.5	0	4.8	36.8	1.1
12	8:7 10:7 1S5	17 00 40	16 40	00	284	250	28	11.4	40.08	36.54	0	109	0.352	0	33.0	0	4.0	11.4	2.4
13	8:7 11:4 1S1	17 40 4	17 00	00	422	230	47	1.6	38.07	32.88	0	126	0.860	0	25.3	0	4.0	17.0	2.0
14	8:7 12:5 1S8	17 10 04	17 00	00	360	155	51	1.3	36.07	15.83	0	99.3	0.532	0	47.9	0	4.8	20.4	2.4
15	8:8 13:0 1S3	80 00 12	51 12	00	214	100	20	4.5	38.07	1.21	0	31.7	0.218	0	16.9	0	4.4	4.8	1.2
16	8:7 14:7 1S5	50 30 20	16 20	00	368	160	71	0.9	50.06	20.70	0	45.3	0.402	0	8.4	0	4.4	20.2	1.2
17	8:7 15:7 1S0	80 00 2	18 11	00	400	290	18	14.1	20.04	3.65	0	13.3	0.700	0	18.2	0	4.4	14.8	2.8
18	8:7 16:5 1S4	60 30 84	20 58	00	420	205	64	2.2	24.04	7.30	0	13.3	0.486	0	34.5	0	4.4	16.0	2.0
19	8:7 17:4 1S7	17 10 04	18 74	00	324	180	74	4.0	36.07	21.92	0	11.5	0.652	0	29.8	0	4.0	22.8	2.8

20	8:18:15 S	7:10:10	1:04:00	000	412	135	20	7	34.06	12.18	0	44.4	1.25	0	49.6	0	4.8	3.2	2.8
21	8:19:18 D	7:10:20	1:04:00	000	220	195	28	6.5	32.06	28.01	0	10.4	0.465	0	35.4	0	4.0	0	2.0
22	8:20:12 D	7:10:29	1:04:48	000	246	115	04	4.3	52.10	3.65	0	38.3	0.288	0	46.1	0	4.4	0	2.0
23	8:20:26 D	7:11:06	1:07:04	000	204	240	95	2	36.07	36.54	0	42.3	0.338	0	21.6	0	4.8	4.8	2.4
24	8:21:14 S	7:12:04	1:09:06	000	280	335	190	6.2	40.08	57.24	0	28.0	0.421	0	63.0	0	2.0	2.4	2.4
25	8:22:11 S	7:16:01	1:07:28	000	446	230	184	14.8	20.04	43.84	0	64.0	0.756	0	82.0	0	2.0	0	2.4
26	8:23:11 S	7:19:04	1:08:56	000	230	205	145	8.32	62.08	24.36	0	75.6	0.643	0	88.4	0	2.0	0	2.4
27	8:24:15 S	7:19:05	1:09:00	000	216	265	184	10.2	40.08	40.19	0	32.5	1.36	0	65.0	0	2.0	0	1.6
	1:10 S	0:00	0:00														8		
28	8:25:11 S	7:20:07	1:02:08	001	552	310	151	6.9	10.02	69.42	0	62.8	1.13	0	10.20	0	2.0	6.8	2.4

Table 2.2 Pedavegi Mandal ----- Winter Season

S. No.	Sample No	pH	EC	TDS	Turbidity	Alkalinity	Hardness	Sodium	Potassium	Calcium	Magnesium	Iron	Chloride	Fluoride	Nitrite	Sulphate	Phosphate	DO	COD	BOD
01	8:11:18 S	8:14:00	1:30:00	1:09:00	0	390	235	76	14.2	72.14	13.29	0	25.5	0.428	00	71	000	5.6	2.4	1.2
02	8:22:18 S	8:17:00	1:33:00	1:02:00	0	456	345	140	2.0	56.16	22.08	0	40.7	0.476	00	84	000	4.6	2.7	2.8
03	8:33:18 S	8:16:05	1:04:40	1:04:00	0	320	170	79	11	33.19	22.12	0	43.7	0.605	00	38	000	4.4	4.8	2.2
04	8:44:18 S	8:20:08	1:00:09	1:08:00	0	456	425	66	5.8	96.12	26.08	0	20.2	1.26	00	40	000	4.4	4.8	2.6
05	8:49:28 S	7:28:00	1:05:00	1:08:00	0	280	235	56	4.2	74.12	11.12	0	16.0	0.589	00	42	000	4.4	1.8	2.6
06	8:50:22 S	8:00:00	1:06:09	1:09:00	0	256	220	48	4.8	40.05	62.15	0	92.5	1.25	00	54	000	4.0	1.6	2.4

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07	8:06:1S	8:52	9:00	690	0	264	220	85	11.2	70.14	10.96	0	298	1.11	000	84	000	44	24	1.8
08	8:07:1S	8:38	1:03	1340	0	566	290	48	3.2	82.16	20.20	0	658	1.06	000	40	000	38	27	1.6
09	8:08:1S	8:44	1:02	850	0	432	230	45	6.4	74.14	10.96	0	400	0.722	000	26	000	36	16	1.8
10	8:09:1S	8:28	1:04	1090	0	538	360	45	2.4	40.05	62.12	0	270	1.99	000	56	000	36	41	2.8
11	8:09:2S	8:50	7:05	990	0	410	245	36	8.9	58.13	21.26	0	103	1.77	000	2	000	50	22	2.0
12	8:10:1S	8:16	8:01	710	0	300	230	45	4.9	58.16	11.12	0	171	0.562	000	40	000	48	24	2.2
13	8:11	8:00	8:07	200	0	340	280	32	12.2	62.18	22.12	0	149	1.29	000	34	000	47	17	2.0
14	8:12:1S	8:00	8:07	700	0	382	305	32	16.4	56.12	11.18	0	159	0.915	000	29	000	42	24	2.4
15	8:13:1S	8:00	9:07	600	0	290	255	83	7.5	59.26	21.26	0	183	0.458	000	30	000	44	20	2.4
16	8:14:1S	8:22	5:03	500	0	188	125	91	9.8	23.06	10.96	0	453	0.626	000	38	000	46	24	2.2
17	8:15:1S	8:44	8:06	700	0	414	205	24	4.4	45.06	11.26	0	725	1.05	000	6	000	44	30	2.0
18	8:16:1S	8:50	8:06	900	0	398	195	43	10.1	33.12	12.08	0	130	0.679	000	18	000	42	14	2.4
19	8:17:1S	8:40	8:07	200	0	318	210	28	10.6	46.12	20.20	0	218	0.760	000	34	000	46	27	2.0
20	8:18:1S	8:40	9:07	800	0	378	230	13	11.2	34.62	24.16	0	151	0.695	000	46	000	42	24	2.0
21	8:19:1D	8:30	6:04	700	0	258	180	86	8.6	24.06	22.06	0	697	1.09	000	22	000	42	18	2.2
22	8:20:1D	7:09	8:06	800	0	242	245	180	9.8	46.08	26.08	0	142	1.97	000	10	000	42	02	2.2
23	8:20:2D	8:01	6:04	500	0	110	170	99	6.4	22.18	16.08	0	102	0.904	000	24	000	40	00	2.2

24	8:21:15S	8:1700	2:500	1:706	0	706	300	163	11.6	64.12	34.10	0	897	0.915	000	76	000	4.4	2.4	2.4
25	8:22:15S	8:1500	2:500	1:706	0	540	325	186	10.4	78.15	31.66	0	838	0.897	000	102	000	4.4	2.2	2.6
26	8:23:16S	8:1006	2:507	1:706	0	486	375	146	11.3	96.26	12.63	0	934	0.706	000	94	000	5.6	2.7	2.8
27	8:24:16S	8:1000	1:400	1:100	0	484	300	133	9.6	48.09	43.89	0	303	1.80	000	46	000	5.6	2.8	2.6
																		6.8		
28	8:25:18S	8:1608	2:507	1:706	0	712	365	200	8.6	60.12	52.37	0	688	1.00	000	110	000	2.4	2.8	1.2
29	8:26:19S	8:1409	1:703	1:102	0	372	325	196	8.3	56.11	45.06	0	558	0.401	000	114	000	3.6	3.3	1.8

Table 2.3 Pedavegi Mandal Summer season

S. No.	Sample No	pH	EC	TDS	Turbidity	Alkalinity	Hardness	Sodium	Potassium	Calcium	Magnesium	Iron	Chloride	Fluoride	Nitrate	Sulphate	Phosphate	DO	CO ₂	BOD
01	8:1:15S	7.62	1.04	1.02	0	326	250	114	10.3	56.92	14.68	0	372	0.421	0	72	0.060	4.0	0	2.4
02	8:2:16S	7.08	2.08	1.08	0	378	260	132	13.4	62.18	22.08	0	418	0.399	0	85	0.068	4.8	0	3.2
03	8:3:16S	7.43	1.08	1.08	0	266	250	103	8.0	46.09	22.12	0	453	0.486	0	38	0.060	4.4	0	2.0
04	8:4:17S	7.37	1.03	1.03	0	320	225	82	3.0	39.67	11.12	0	373	0.970	0	41	0.077	4.0	3.2	2.0
05	8:4:19S	7.08	1.06	1.06	0	280	240	68	4.0	60.12	9.62	0	96	1.05	0	46	0.01	4.4	0	2.0

06	8:5:1S	7:63	1:40	8:96	0	266	250	77	2.9	48.06	1.218	0	549	0.876	0.011	83	0.100	4.0	8.0	2.0
07	8:6:1S	7:77	9:00	5:76	0	308	165	75	4.9	52.19	1.612	0	492	1.00	0	40	0.082	4.8	1.6	2.4
08	8:7:1S	7:45	1:20	7:68	0	250	195	84	4.6	60.12	12.19	0	530	0.873	0	27	0.105	4.4	0.0	2.4
09	8:8:1S	7:70	1:40	8:96	0	320	230	114	4.0	58.16	11.12	0	512	0.806	0	56	0.073	4.0	0.0	2.0
10	8:9:1S	7:97	1:30	8:32	0	282	160	54	2.8	40.08	10.12	0	452	1.92	0	2	0.055	4.0	0.0	1.6
11	8:9:2S	7:79	1:50	9:60	0	480	305	92	4.6	56.12	11.18	0	510	2.06	0	41	0.086	4.8	0.0	2.4
12	8:10:1S	7:95	1:00	6:40	0	220	210	92	14.5	60.18	9.62	0	185	0.484	0	35	0.064	4.0	1.6	2.0
13	8:11:1S	7:73	1:18	8:30	0	300	250	61	2.4	58.13	21.26	0	146	1.00	0	29	0.105	4.0	0.0	2.4
	8:11:1S	7:48	1:00	2:00																8
14	8:12:1S	7:57	1:10	7:04	1	290	215	64	1.2	28.12	26.16	0	160	0.806	0	45	0.091	4.0	0.0	2.0
15	8:13:1S	7:62	1:10	7:04	0	222	200	94	12.8	58.12	22.15	0	201	0.361	0	38	0.064	4.0	0.0	2.0
16	8:14:1S	7:90	1:00	4:48	0	204	175	54	0.6	60.18	13.18	0	545	0.486	0.018	6	0.043	4.0	0.0	2.0
17	8:15:1S	7:80	1:00	6:40	1	330	175	95	14	59.16	12.12	0	805	0.924	0.023	18	0.077	3.8	0.0	2.4
18	8:16:1S	7:90	1:00	6:40	1	448	190	94	2.5	42.18	13.16	0	103	0.609	0.029	34	0.055	3.2	0.0	1.6
19	8:17:1S	7:60	1:10	7:04	1	380	150	96	7.1	46.19	12.12	0	220	0.920	0	46	0.035	4.0	0.0	2.0
20	8:18:1S	7:60	1:20	7:08	0	210	240	94	11.6	32.16	10.15	0	176	1.65	0	22	0.067	4.8	0.0	2.4

21	8:19 :1 D	8 0 0 7	7 0 0 4	4 4 8	0	252	20 0	45	3.3	24. 16	12.9 8	0	89. 6	1.2 5	0	10	0.0 37	4 8	0	2. 4
22	8:20 :1 D	7 6 8	1 0 0	6 4 0	0	196	16 0	64	4.9	34. 12	6.98	0	74. 6	0.3 94	0	24	0.0 37	4 8	0	2. 4
23	8:20 :2 D	7 8 7	7 0 4	4 4 8	0	312	20 0	45	2.5	34. 68	5.12	0	43. 4	0.4 55	0	76	0.0 45	4 4	0	2. 0
24	8:21 :1 S	7 7 6	2 7 0	1 7 2	1	436	31 0	14 5	9.0	30. 12	10.1 8	0	64. 6	0.5 11	0	10 2	0.0 63	4 0	0	2. 0
25	8:22 :1 S	7 7 0	2 7 2	1 7 2	1	512	36 5	16 0	12. 4	46. 18	11.1 2	0	69. 0	0.9 66	0	94	0.0 56	4 8	0	2. 4
26	8:23 :1 S	7 3 9	2 6 0	1 6 6	1	252	16 5	13 6	10. 4	52. 16	10.1 2	0	65. 8	0.8 76	0	46	0.0 56	4 0	1. 6	2. 0
27	8:24 :1 S	7 5	1 6	9 4	1	460	31 0	11 2	10. 2	60. 12	12.0 2	0	32. 2	1.1 5	0	52	0.0 1	4 8	0	2. 0
	:1 0	9 0	0 0	0														0		
28	8:25 :1 S	7 7 8	3 0 0	2 1 2	1	560	39 0	15 2	7.2	41. 52	10.1 2	0	82. 5	1.0 5	0	11 0	0.0 95	3 6	6. 4	1. 6
29	8:26 :1 S	7 1 5	2 3 0	1 3 4	1	352	21 5	12 2	4.1	40. 08	21.1 8	0	56. 0	0.5 55	0	11 4	0.0 59	4 0	2. 8	2. 0

II. Results And Discussion

The physicochemical parameters of various water samples analyzed were presented in Table -2 and the results are compared with standard limits prescribed by WHO.

p^H

p^H of the water samples analyzed for the present study was within WHO limits and all of them are slightly alkaline in nature. Water which has P^H Value of more than 9 or less than 4.5 becomes unsuitable for Use of drinking⁸. Most of the samples analyzed for P^H are within the limits of WHO standards. Few samples show slightly higher the range due to presence of the dissolved solids.

Electrical conductivity (E.C)

Electrical conductivity is a measure of water capacity to convey electrical current. It signifies the amount of total dissolved solids^{9, 10}. All the samples (1-29) are within the range in three seasons except sample no 26(2900) in rainy season and sample no 28 in summer season(3300), Which shows slightly higher the range due to presence of increasing the ions .

Total dissolved solids (T D S).

TDS is an important parameter which imparts a particular taste to water and reduce its potability. The permissible range of TDS for drinking water is 500 mg/L. All the samples (1-29) are within the range in three seasons except sample no 28 in summer season(2112).This shows slightly higher the range due to increasing the dissolved solids .High concentration ground water are generally not harmful to the human being but high concentrations of these may affect persons who are suffering from kidney and heart diseases¹¹.

Total Alkalinity (T.A)

The value of alkalinity in water provides an idea of natural salts present in water. The standard desirable limit of alkalinity for potable water is 120 mg / L. The maximum permissible limit is 600 mg/L. Higher alkalinity gives unpleasant taste to water. All the samples (1-29) are within the range in three seasons; except the sample no 24 & 28 having the value is 706 & 712 is due to presence of carbonates and hydroxides. Alkalinity in itself is not harmful to human being, but water samples with less than 100 mg/L are desirable for domestic use¹².

Total Hardness (T H)

Hardness is a property of water, which does not produce foam or leather freely when treated with soap solution. It is mainly due to the presence of calcium or Magnesium salts or both. Total hardness of the water samples are varied from 200 to 600. Hardness of all the water samples analyzed is within the limits of WHO standards. Excess hardness in the water leads to heart diseases and kidney stone formation¹³.

Calcium (Ca) and Magnesium (Mg):

The main source of calcium in ground water is leaching of rocks. It plays an important role in the formation of bones. Excess of calcium than the permeable limits causes gastrointestinal diseases and stone formation¹⁴. In ground water, generally Magnesium content will be less than Calcium content. More than the permeable range of Magnesium leads to unpleasant taste to water. Calcium and Magnesium concentrations analyzed for all drinking water samples are within the limits of WHO standards (Ca 75 mg/L and for Mg 50 mg / L). All the samples (1-29) are within the range in three seasons.

Dissolved Oxygen (D.O) and Biological Oxygen demand (B.O.D)

D.O is essential for aquatic life. A low D.O (Less than 2 mg/L) would indicate poor water quality and this would cause sustainability of aquatic life is difficult. B.O.D is a measure of organic material contamination in water. It indicates the amount of oxygen required for oxidation organic impurities as well as some inorganic materials like Sulphites etc. All the samples (1-29) analyzed for D.O and B.O.D are within the permissible range in all the three seasons

Fluoride

The main source of fluoride in water is, leaching of fluoride containing minerals in to the ground water as the rain water percolates through the earth. Excess of intake of fluoride through drinking water causes dental fluorosis, mild skeletal fluorosis¹⁵. In the present analysis Fluoride concentration was found to be varied from 1.218 to 2.06. For few samples the values are more than the permeable limits of WHO standards (1.0 to 1.5 mg/L). Soil – water – rock inters actions play an important role in this regard.

Sodium and Potassium

All natural waters contain some sodium. Sodium concentrations above 200 mg/L may alter the taste of water. High levels of sodium in drinking-water are associated with increased blood pressure. According to WHO standards, concentration of sodium in drinking water is 200 mg/l. All the samples of water analyzed for drinking are within the permissible limits of WHO limits.

Potassium is an essential element for human nutrition. According to WHO standards the permissible limit of potassium is 12 mg/l. Most of the water samples analyze d for drinking are within the permissible limits. Very Few samples show slight excess of WHO limit. Concentrations of potassium normally found in drinking-water are generally low and slight excess does not pose any health problems.

Chlorides

Chlorides are usually present in water. Presence of chlorides in water above the permissible limit is an indicator of pollution. High concentrations of chlorides have no adverse effects to human being, but it gives laxative effect.¹⁶. The permissible limits of chlorides for drinking water is 500 mg/L. The present study indicates that the concentration of chlorides in all the samples is within the permissible limits.

Sulphate

Sulphate is found in small quantities in ground water. Sulphate may come into bore water by industrial or anthropogenic additions in form of fertilizers. All the samples analyzed for sulphate concentration are within the permissible limits of WHO standards. High concentrations of Sulphate cause Laxative effect to he children in hot weather climates¹⁷.

Nitrate

Groundwater contains nitrate due to leaching of nitrate with the percolating water. Groundwater can also be contaminated by sewage and other wastes rich in nitrates. The tolerance range for nitrate is 20 mg/L to 45 mg/L. Higher levels of nitrate in the drinking water source may be due to the excess usage of fertilizers and pesticides by the people residing in this area. The nitrate content in the study area varied in the range 0.041 mg/L to 0.75 mg/L and found within the prescribed limit. Excessive concentrations of Nitrates in drinking water causes Methemoglobinaemia¹⁸.

Conclusion

In the present study, bore water samples are collected from twenty eight different villages of Pedavegi mandal. The water samples are analyzed for various physico chemical parameters like P^H, TH, TDS, Alkalinity, EC, calcium, Magnesium, sodium, potassium, Chlorides, Nitrates, Sulphates, DO and BOD. The results are compared with WHO standards for drinking purpose. The result reveals that all the sources of bore water in the study area are suitable for drinking purpose, yet it needs few treatments to minimize some contaminations especially total hardness and fluoride which are reported to be higher than WHO standards.

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